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## Structure Reports

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## Bis[glycinium(0.5+)] perrhenate

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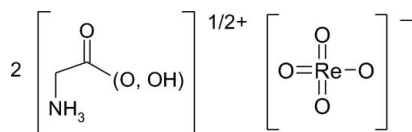
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Key indicators: single-crystal X-ray study;  $T = 291$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å; disorder in solvent or counterion;  $R$  factor = 0.030;  $wR$  factor = 0.055; data-to-parameter ratio = 60.9.

All the residues of the title compound,  $(\text{C}_2\text{H}_5.5\text{NO}_2)_2[\text{ReO}_4]$ , are located in general crystallographic positions. The glycine molecules have usual conformations [Rodrigues Matos Beja *et al.* (2006). *Acta Cryst. C* **62**, o71–o72] with the H atom of the carboxylate group half-occupied, thus bearing a formal half-positive charge per molecule. The perrhenate anion has nearly ideal tetrahedral geometry. A large number of strong hydrogen bonds give rise to the overall three-dimensional network. A two-dimensional network, parallel to (100), is made up of strong  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonds with a donor acceptor distance of 2.445 (2) Å. A large number of weaker  $\text{O}-\text{H}\cdots\text{O}$  and  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bonds consolidates the structure into an overall three-dimensional network.

## Related literature

For a related structure, see: Rodrigues *et al.* (2006).

## Experimental

## Crystal data

$(\text{C}_2\text{H}_5.5\text{NO}_2)_2[\text{ReO}_4]$   
 $M_r = 401.35$   
 Monoclinic,  $P2_1/c$   
 $a = 15.7095$  (5) Å  
 $b = 8.1826$  (3) Å

$c = 8.2909$  (3) Å  
 $\beta = 103.7152$  (16)°  
 $V = 1035.36$  (6) Å<sup>3</sup>  
 $Z = 4$   
 Mo  $K\alpha$  radiation

$\mu = 11.77$  mm<sup>-1</sup>  
 $T = 291$  (2) K

0.15 × 0.13 × 0.10 mm

## Data collection

Bruker APEXII diffractometer  
 Absorption correction: multi-scan  
 (*SADABS*; Sheldrick, 2003)  
 $T_{\min} = 0.18$ ,  $T_{\max} = 0.31$

78826 measured reflections  
 8587 independent reflections  
 6232 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.038$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.030$   
 $wR(F^2) = 0.055$   
 $S = 1.06$   
 8587 reflections

141 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\max} = 2.33$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -2.87$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{O12}-\text{H12}\cdots\text{O22}^{\text{i}}$	0.82	1.64	2.445 (2)	167
$\text{N11}-\text{H11A}\cdots\text{O11}^{\text{ii}}$	0.89	2.02	2.869 (2)	158
$\text{N11}-\text{H11B}\cdots\text{O21}$	0.89	2.18	3.003 (2)	153
$\text{N11}-\text{H11B}\cdots\text{O2}$	0.89	2.47	3.000 (3)	119
$\text{N11}-\text{H11C}\cdots\text{O11}^{\text{iii}}$	0.89	1.94	2.830 (3)	175
$\text{O22}-\text{H22}\cdots\text{O12}^{\text{i}}$	0.82	1.64	2.445 (2)	165
$\text{N21}-\text{H21A}\cdots\text{O21}$	0.89	2.27	2.738 (3)	113
$\text{N21}-\text{H21A}\cdots\text{O1}$	0.89	2.29	3.136 (4)	158
$\text{N21}-\text{H21B}\cdots\text{O3}^{\text{iv}}$	0.89	2.10	2.896 (3)	149
$\text{N21}-\text{H21B}\cdots\text{O1}^{\text{v}}$	0.89	2.60	3.274 (4)	133
$\text{N21}-\text{H21C}\cdots\text{O4}^{\text{vi}}$	0.89	2.14	2.794 (3)	130
$\text{N21}-\text{H21C}\cdots\text{O1}^{\text{vii}}$	0.89	2.37	3.012 (3)	130

Symmetry codes: (i)  $-x, -y + 1, -z$ ; (ii)  $-x, y - \frac{1}{2}, -z + \frac{1}{2}$ ; (iii)  $x, -y + \frac{3}{2}, z - \frac{1}{2}$ ; (iv)  $-x + 1, -y + 1, -z$ ; (v)  $x, -y + \frac{1}{2}, z - \frac{1}{2}$ ; (vi)  $x, -y + \frac{1}{2}, z + \frac{1}{2}$ ; (vii)  $-x + 1, y - \frac{1}{2}, -z + \frac{1}{2}$ .

Data collection: *APEX2* (Bruker–Nonius, 2004); cell refinement: *SAINT* (Bruker, 2003); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEPII* (Johnson, 1976); software used to prepare material for publication: *SHELXL97* and *PLATON* (Spek, 2003).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: RN2049).

## References

- Bruker (2003). *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.  
 Bruker (2004). *APEX2*. Bruker AXS Inc., Madison, Wisconsin, USA.  
 Johnson, C. K. (1976). *ORTEPII*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.  
 Rodrigues, V. H., Matos Beja, A., Paixão, J. A. & Costa, M. M. R. R. (2006). *Acta Cryst. C* **62**, o71–o72.  
 Sheldrick, G. M. (2003). *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.  
 Sheldrick, G. M. (2008). *Acta Cryst. A* **64**, 112–122.  
 Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.

**supplementary materials**

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## Bis[glycinium(0.5+)] perrhenate

V. H. Rodrigues, M. M. R. R. Costa, T. Dekola and E. de Matos Gomes

### Experimental

Crystals of the monoclinic polymorph of glycinium glycine perrhenate were obtained from a water solution of analytical grade reagents glycine(99.5%) and perrhenic acid solution (65–75% water, 99.5%), purchased from Aldrich, in a 2:1 molar ratio.

### Refinement

The structure was solved by direct methods using *SHELXS97*. All H atoms were first located on a difference Fourier map; those bonded to C atoms and carboxyl O atoms were placed at idealized positions and refined as riding [C—H=0.97 and 0.98 Å, O—H=0.82 Å,  $U_{\text{iso}}(\text{H})=1.2U_{\text{eq}}(\text{C})$  and  $U_{\text{iso}}(\text{H})=1.5U_{\text{eq}}(\text{O})$ ].

Examination of the crystal structure with *PLATON* (Spek, 2003) showed that there are no solvent-accessible voids in the crystal lattice.

### Figures

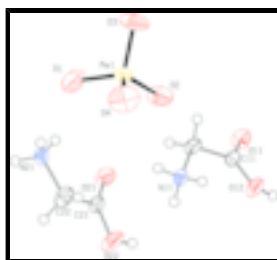


Fig. 1. *ORTEP* (Johnson, 1976) plot of the title compound. Displacement ellipsoids are drawn at the 50% level.

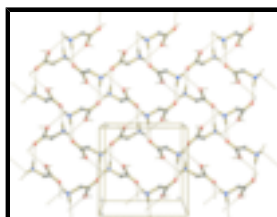


Fig. 2. Packing of glycine molecules showing the (100) network.

## Bis[glycinium(0.5+)] perrhenate

### Crystal data

$2(\text{C}_2\text{H}_5.5\text{NO}_2)[\text{ReO}_4]$

$M_r = 401.35$

Monoclinic,  $P2_1/c$

$F_{000} = 752$

$D_x = 2.575 \text{ Mg m}^{-3}$

Mo  $K\alpha$  radiation

$\lambda = 0.71073 \text{ \AA}$

# supplementary materials

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Hall symbol: -P 2ybc  
 $a = 15.7095$  (5) Å  
 $b = 8.1826$  (3) Å  
 $c = 8.2909$  (3) Å  
 $\beta = 103.7152$  (16)°  
 $V = 1035.36$  (6) Å<sup>3</sup>  
 $Z = 4$

Cell parameters from 7839 reflections  
 $\theta = 4.0$ – $40.1$ °  
 $\mu = 11.77$  mm<sup>-1</sup>  
 $T = 291$  (2) K  
Block, translucent colourless  
 $0.15 \times 0.13 \times 0.10$  mm

## Data collection

Bruker APEXII  
diffractometer  
Radiation source: fine-focus sealed tube  
Monochromator: graphite  
 $T = 291$ (2) K  
 $\varphi$  and  $\omega$  scans  
Absorption correction: multi-scan  
(SADABS; Sheldrick, 2003)  
 $T_{\min} = 0.18$ ,  $T_{\max} = 0.31$   
78826 measured reflections

8587 independent reflections  
6232 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.038$   
 $\theta_{\text{max}} = 45.3$ °  
 $\theta_{\text{min}} = 2.7$ °  
 $h = -30 \rightarrow 31$   
 $k = -16 \rightarrow 15$   
 $l = -16 \rightarrow 16$

## Refinement

Refinement on  $F^2$   
Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.030$   
 $wR(F^2) = 0.055$   
 $S = 1.06$   
8587 reflections  
141 parameters  
Primary atom site location: structure-invariant direct methods  
Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites  
H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.001P)^2 + 2.2865P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} = 0.002$   
 $\Delta\rho_{\text{max}} = 2.33$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -2.87$  e Å<sup>-3</sup>  
Extinction correction: SHELXL97 (Sheldrick, 2008),  
 $F_c^* = kFc[1 + 0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$   
Extinction coefficient: 0.00614 (15)

## Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
Re1	0.384633 (5)	0.604894 (11)	0.053166 (13)	0.02896 (3)	
O1	0.43021 (16)	0.4923 (4)	0.2269 (3)	0.0636 (7)	
O2	0.28327 (14)	0.6733 (3)	0.0625 (4)	0.0633 (8)	
O3	0.44936 (16)	0.7699 (3)	0.0397 (5)	0.0685 (9)	
O4	0.3784 (2)	0.4775 (4)	-0.1121 (3)	0.0698 (8)	
O11	0.02851 (12)	0.9119 (2)	0.2663 (2)	0.0343 (4)	
O12	-0.02823 (11)	0.7477 (2)	0.0538 (2)	0.0366 (4)	
H12	-0.0747	0.7890	0.0614	0.055*	0.50
C11	0.03285 (13)	0.7965 (3)	0.1730 (3)	0.0242 (3)	
C12	0.11921 (13)	0.7067 (3)	0.1971 (3)	0.0291 (4)	
H12A	0.1608	0.7754	0.1588	0.035*	
H12B	0.1424	0.6864	0.3145	0.035*	
N11	0.11042 (11)	0.5501 (2)	0.1070 (2)	0.0253 (3)	
H11A	0.0761	0.4833	0.1479	0.038*	
H11B	0.1631	0.5048	0.1189	0.038*	
H11C	0.0868	0.5676	-0.0002	0.038*	
O21	0.24920 (12)	0.3008 (2)	0.0961 (3)	0.0409 (4)	
O22	0.16419 (12)	0.1024 (3)	-0.0423 (3)	0.0504 (6)	
H22	0.1241	0.1641	-0.0353	0.076*	0.50
C21	0.23754 (14)	0.1710 (3)	0.0210 (3)	0.0281 (4)	
C22	0.31325 (16)	0.0716 (3)	-0.0074 (4)	0.0373 (5)	
H22A	0.3076	-0.0399	0.0286	0.045*	
H22B	0.3109	0.0690	-0.1254	0.045*	
N21	0.39933 (12)	0.1372 (3)	0.0823 (3)	0.0337 (4)	
H21A	0.3923	0.2372	0.1194	0.051*	
H21B	0.4351	0.1417	0.0136	0.051*	
H21C	0.4224	0.0724	0.1676	0.051*	

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Re1	0.02220 (4)	0.02717 (4)	0.04083 (5)	-0.00013 (3)	0.01407 (3)	-0.00404 (3)
O1	0.0401 (12)	0.093 (2)	0.0524 (14)	-0.0127 (13)	-0.0001 (10)	0.0177 (14)
O2	0.0333 (10)	0.0379 (10)	0.130 (2)	0.0040 (8)	0.0416 (13)	-0.0007 (13)
O3	0.0467 (12)	0.0383 (11)	0.136 (3)	-0.0090 (9)	0.0521 (16)	-0.0019 (14)
O4	0.080 (2)	0.083 (2)	0.0523 (15)	-0.0094 (16)	0.0276 (14)	-0.0268 (14)
O11	0.0309 (7)	0.0398 (9)	0.0320 (8)	0.0054 (7)	0.0069 (6)	-0.0106 (7)
O12	0.0226 (7)	0.0378 (9)	0.0440 (10)	0.0077 (6)	-0.0031 (7)	-0.0144 (7)
C11	0.0206 (7)	0.0270 (8)	0.0257 (9)	0.0011 (6)	0.0068 (6)	-0.0006 (7)
C12	0.0194 (7)	0.0314 (9)	0.0348 (11)	0.0015 (7)	0.0029 (7)	-0.0069 (8)
N11	0.0187 (6)	0.0262 (7)	0.0300 (9)	0.0035 (5)	0.0043 (6)	0.0002 (6)
O21	0.0259 (7)	0.0310 (8)	0.0635 (13)	0.0041 (6)	0.0063 (8)	-0.0144 (8)
O22	0.0219 (7)	0.0475 (11)	0.0747 (15)	0.0057 (7)	-0.0029 (8)	-0.0284 (11)
C21	0.0213 (8)	0.0288 (9)	0.0323 (10)	0.0054 (7)	0.0027 (7)	-0.0034 (8)

## supplementary materials

C22	0.0254 (9)	0.0429 (13)	0.0417 (13)	0.0092 (9)	0.0043 (9)	-0.0133 (10)
N21	0.0217 (7)	0.0309 (9)	0.0490 (12)	0.0096 (6)	0.0094 (8)	0.0134 (8)

### Geometric parameters (Å, °)

Re1—O4	1.706 (3)	N11—H11B	0.8900
Re1—O2	1.707 (2)	N11—H11C	0.8900
Re1—O3	1.710 (2)	O21—C21	1.223 (3)
Re1—O1	1.717 (3)	O22—C21	1.277 (3)
O11—C11	1.233 (3)	O22—H22	0.8200
O12—C11	1.267 (3)	C21—C22	1.505 (3)
O12—H12	0.8200	C22—N21	1.481 (3)
C11—C12	1.514 (3)	C22—H22A	0.9700
C12—N11	1.473 (3)	C22—H22B	0.9700
C12—H12A	0.9700	N21—H21A	0.8900
C12—H12B	0.9700	N21—H21B	0.8900
N11—H11A	0.8900	N21—H21C	0.8900
O4—Re1—O2	111.10 (15)	C12—N11—H11C	109.5
O4—Re1—O3	110.60 (15)	H11A—N11—H11C	109.5
O2—Re1—O3	108.64 (11)	H11B—N11—H11C	109.5
O4—Re1—O1	106.21 (17)	C21—O22—H22	109.5
O2—Re1—O1	110.24 (14)	O21—C21—O22	127.1 (2)
O3—Re1—O1	110.03 (15)	O21—C21—C22	121.4 (2)
C11—O12—H12	109.5	O22—C21—C22	111.5 (2)
O11—C11—O12	125.87 (19)	N21—C22—C21	112.7 (2)
O11—C11—C12	118.03 (19)	N21—C22—H22A	109.0
O12—C11—C12	116.08 (18)	C21—C22—H22A	109.0
N11—C12—C11	112.44 (17)	N21—C22—H22B	109.0
N11—C12—H12A	109.1	C21—C22—H22B	109.0
C11—C12—H12A	109.1	H22A—C22—H22B	107.8
N11—C12—H12B	109.1	C22—N21—H21A	109.5
C11—C12—H12B	109.1	C22—N21—H21B	109.5
H12A—C12—H12B	107.8	H21A—N21—H21B	109.5
C12—N11—H11A	109.5	C22—N21—H21C	109.5
C12—N11—H11B	109.5	H21A—N21—H21C	109.5
H11A—N11—H11B	109.5	H21B—N21—H21C	109.5
O11—C11—C12—N11	166.3 (2)	O21—C21—C22—N21	7.3 (4)
O12—C11—C12—N11	-15.7 (3)	O22—C21—C22—N21	-172.9 (2)

### Hydrogen-bond geometry (Å, °)

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
O12—H12 $\cdots$ O22 <sup>i</sup>	0.82	1.64	2.445 (2)	167
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N21—H21B···O1 <sup>v</sup>	0.89	2.60	3.274 (4)	133
N21—H21C···O4 <sup>vi</sup>	0.89	2.14	2.794 (3)	130
N21—H21C···O1 <sup>vii</sup>	0.89	2.37	3.012 (3)	130

Symmetry codes: (i)  $-x, -y+1, -z$ ; (ii)  $-x, y-1/2, -z+1/2$ ; (iii)  $x, -y+3/2, z-1/2$ ; (iv)  $-x+1, -y+1, -z$ ; (v)  $x, -y+1/2, z-1/2$ ; (vi)  $x, -y+1/2, z+1/2$ ; (vii)  $-x+1, y-1/2, -z+1/2$ .

Fig. 1

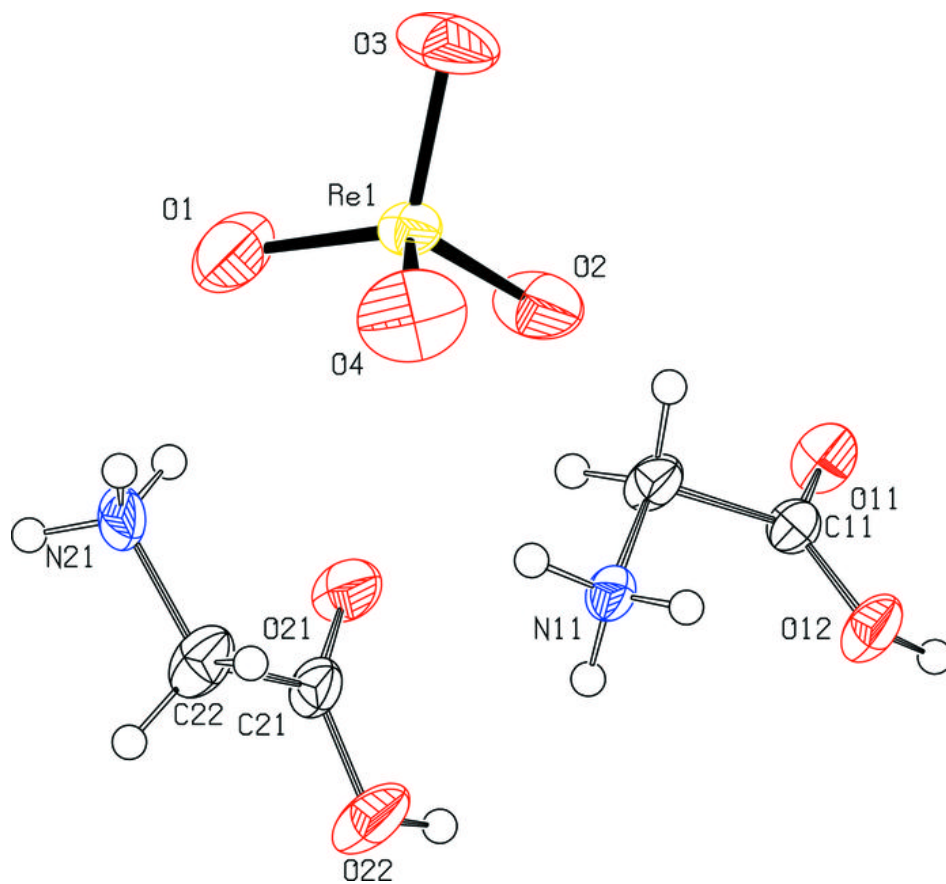




Fig. 2

